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magnetic hyperfine interaction in free atoms

or

coupling of angular momenta : from L-S to I-J **COUPLING Of

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coupling of angular momenta: L-S

We'll remind first what you saw in earlier courses on the coupling of orbital and spin angular momenta in an atom:

stefaan.cottenier@ugent.be 1 The problem: "For a given shell (n,l), how do a given number of electrons occupy the available orbitals?"

Example: C (n=2, l=1), 2 p-electrons

There are 6 different cricitals: (m;--1,0,+1,
and this for either spin), hence 6x6=36 possibilities
to put these 2 electrons. Which of those 36
possibilities has the lowest energy (and will
therefore be found as the ground

Hund's rules provide you with an algorithm to find this ground state

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coupling of angular momenta: L-S

1 st Hund's rule

Only configurations where the total S is maximal should be considered further.

 S is found as the absolute value of the sum of $all \, m_S$ values

Our example: only states with S=1 (twice m_S=+1/2)
or twice m_S=-1/2) should be considered further.
 =+1/2

2nd Hund's rule

Within the previous set, only configurations where the total L is maximal should be considered further.

L is found as the absolute value of the sum of $\mathsf{all} \mathsf{\ m}_\mathsf{L}$ values

Our example: states with S=1 cannot contain 2 electrons
in the same m_L orbital. Hence, the maximal L
is L=1 (two electrons in m_L=+1,0, or in m_L=-1,0) 4

coupling of angular momenta: L-S

How many of our 36 states are left if we restrict ourselves to S=1 and L=1? Two ways to count:

First way :

S=1 can have three different orientations (2S+1=3; m_s=-1,0,+1)
L=1 can have three different orientations (2L+1=3; m_L=-1,0,+1)

\rightarrow 3x3=9 out of 36

Second way:

Angular momenta coupling rules: S=1 and L=1 can couple to J=L+S,..,|L-S| = 2,1,0 Each J-value has 2J+1 orientations: 5, 3, 1

 $5\frac{1}{2}$

 \rightarrow 5+3+1=9 out of 36

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These the blends and state and a state called a state and a state and a

There is the blend contain and the state of the state of the state of Which of the remaining (9) states is the ground state? 3 rd Hund's rule Of the remaining states, those with the lowest energy are the ones with • J minimal if the shell is less than half-filled • J maximal if the shell is more than half-filled Physical picture: mutual orientation of L and S
Example: 2 electrons in a p-shell is less than half-filled $\bigcup_{L=1, S=1}$ **••** If there is no interaction between Left We restrict correctes to 5-1 and L-17

• If the way :

S-1 can between different orientations $(26+1-2; m_2-1.0+1)$

• S-2-1 can be these different containings $(24+1-2; m_2-1.0+$ **• If there is interaction (spin-orbit coupling)**, some will have a lower than output \mathbb{R} and $\mathbb{$ $J=0$ 6 and 6 and 7 and J=1 $J=2$ and $J=2$ and Strick m_e = 1.0, + 1)
 $\frac{1}{2}$
 $\frac{1}{2}$ (and couple to $j\pm1$, ts_{0, -1}1, 58 = 2, 1, 0
 $\frac{1}{2}$
 $\frac{1$ $6\qquad \qquad$ coupling of angular momenta: L-S

coupling of angular momenta: I-J nuclear magnetic moment operator $\hat{\mu}_I = \frac{\mu}{I \hbar} \hat{I}$ experimentally known – the 'size' of the nuclear magnetic moment (scalar). Magnetic hyperfine field operator: The perturbing hamiltonian H_1 : : Andrew March 1995, and the Company of th $\hat{H}_{ij} = -\hat{\boldsymbol{\mu}}_I \cdot \hat{\boldsymbol{B}}(\boldsymbol{0})$ $\begin{split} &= -\frac{\mu B_J}{\hbar^2\,IJ}\,\hat{\pmb{I}}\cdot\hat{\pmb{J}} \\ &= -\frac{\mu B_J}{2\hbar^2\,IJ}\,(\hat{F}^2-\hat{I}^2-\hat{J}^2) \end{split}$ Use $\dot{F}^2 = \dot{F}^2 = (\dot{I} + \dot{J})^2 = \dot{I}^2 + \dot{J}^2 + 2\dot{I} \cdot \dot{J}$

coupling of angular momenta: I-J

Apply perturbation theory

The states of the unperturbed system are the $|F>$ (direct product of $||>$ and $|J>$)

The perturbing hamiltonian is likely to lift degeneracies person theory for the degeneration theory for the degenrate case
- \rightarrow perturbation theory for the degenrate case

Fortunately the |F> states are orthonormal already (property of angular momentum eigenstates)

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